

LEWIS ACID-CATALYSED REARRANGEMENT OF STEROIDAL OXETANES:  
REVISION OF THE COURSE OF SOLVOLYSIS OF 3 $\beta$ -TOSYLOXY-5 $\beta$ -CHOLESTAN-5-OL-6-ONE.†

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Summary The BF<sub>3</sub>-catalysed rearrangement of 6 $\beta$ -acetoxy-3 $\alpha$ ,5-epoxy-5 $\alpha$ -cholestane gave the 3 $\alpha$ ,5 $\beta$ -diol, the 3 $\alpha$ ,10 $\alpha$ -epoxide, and the 2 $\alpha$ ,5 $\alpha$ -epoxide, and the product of solvolysis of 3 $\beta$ -tosyloxy-5 $\beta$ -cholestan-5-ol-6-one was identified as 3 $\alpha$ ,5-epoxy-A-homo-B-nor-5 $\alpha$ -cholestan-4a-one.

The reaction of 3 $\alpha$ ,5-epoxy-5 $\alpha$ -cholestane (1) with BF<sub>3</sub>.Et<sub>2</sub>O was reported, by Henbest and co-workers, to give epicholesterol.<sup>1</sup> However, the 3 $\beta$ ,5 $\beta$ -oxetan-6-one (7), which was prepared by solvolysis of 3 $\beta$ -tosyloxy-5 $\beta$ -cholestan-5-ol-6-one (5) was reported to be unreactive towards BF<sub>3</sub>.Et<sub>2</sub>O and the 6 $\alpha$ -acetoxy-3 $\beta$ ,5 $\beta$ -oxetane (8) and its trifluoroacetoxy analogue, which were prepared from (7) by reduction and acylation, were unreactive towards a number of acidic reagents.<sup>2</sup> We have shown<sup>3</sup> that the rearrangements of some steroidal 5,6-epoxides are influenced markedly by electron withdrawing groups at C-3 and we were interested to determine whether the lack of reactivity of the 3 $\beta$ ,5 $\beta$ -oxetanes related to the electron withdrawing groups at C-6 or to the configuration of the 3,5-epoxide bridge as suggested by Rowland.<sup>2</sup> Accordingly, we have investigated the BF<sub>3</sub>-catalysed rearrangement of the 6 $\beta$ -acetoxy-3 $\alpha$ ,5 $\alpha$ -oxetane (2).<sup>4</sup> Our method of preparation of (2) differed slightly from that previously reported.<sup>4</sup> Monoperphthalic acid oxidation of cholesteryl tosylate gave the 5 $\xi$ ,6 $\xi$ -epoxides which were hydrolysed to the 3 $\beta$ -tosyloxy-5 $\alpha$ ,6 $\beta$ -diol (6).<sup>5</sup> Reaction of the compound (6) with Bu<sup>t</sup>-OK/Bu<sup>t</sup>-OH gave the 6 $\beta$ -hydroxy-3 $\alpha$ ,5 $\alpha$ -oxetane (3) which on acetylation gave the required acetate (2).

Reaction of 6 $\beta$ -acetoxy-3 $\alpha$ ,5 $\alpha$ -oxetane (2) in benzene with BF<sub>3</sub>.Et<sub>2</sub>O gave essentially the 6 $\beta$ -acetoxy-3 $\alpha$ ,5 $\beta$ -diol (9) (26%), the 6 $\beta$ -acetoxy-3 $\alpha$ ,10 $\alpha$ -epoxide (12) (16%),<sup>6</sup> and the 6 $\beta$ -acetoxy-2 $\alpha$ ,5 $\alpha$ -epoxide (13) (10%). The structure of the 6 $\beta$ -acetoxy-3 $\alpha$ ,10 $\alpha$ -epoxide (12) was confirmed by its

† Part of this work was reported at the East Midlands Meeting of the Perkin Division of The Chemical Society at Nottingham on December 18th, 1978.

spectroscopic data and its m.p. and mixed m.p. with an authentic sample.<sup>†</sup> The structure of the 6 $\beta$ -acetoxy-3 $\alpha$ ,5 $\beta$ -diol (9) was confirmed by its conversion via the ketol (10) into the known<sup>7</sup> 6 $\beta$ -acetoxycholest-4-en-3-one (15). The compound (15) was also prepared from 6 $\alpha$ -acetoxy-5 $\alpha$ -cholestane-3 $\beta$ ,5-diol via the ketol (11)<sup>8</sup> which was shown to be different from the ketol (10). The structure of the 6 $\beta$ -acetoxy-2 $\alpha$ ,5 $\alpha$ -epoxide (13) was established by comparison with a sample prepared by Pd/C-hydrazine debromination of the 6 $\beta$ -acetoxy-3 $\xi$ -bromo-2 $\alpha$ ,5 $\alpha$ -epoxides (14).<sup>9</sup>

The rearrangement of the 6 $\beta$ -acetoxy-3 $\alpha$ ,5 $\alpha$ -oxetane (2) (Scheme 1) presumably proceeds via cleavage of the C-5-O bond, or, to a lesser extent, the C-3-O bond leading to the C-5 or the C-3 carbonium ions.\*\* The C-5 carbonium ion may rearrange to the C-10 carbonium ion and the 3,10-ether (12) or it may be trapped by the 6 $\beta$ -acetoxy-group leading to the relatively stable acetoxonium ion (6) which, on aqueous work-up, would give the 6 $\beta$ -acetoxy-3 $\alpha$ ,5 $\beta$ -diol (9). The C-3 carbonium ion could rearrange by a hydride shift to the C-2 carbonium ion and the 2,5-ether (13). The intermediacy of the acetoxonium ion (16) was supported by isolation of the dioxolane (17) when the rearrangement was carried out in ether in the presence of NaBH<sub>4</sub>. The <sup>1</sup>H n.m.r. spectrum of the dioxolane (17) showed characteristic signals at  $\delta$ 5.12 (q, MeCH<sub>2</sub>O<sub>2</sub>) and  $\delta$ 1.43 (d, MeCH<sub>2</sub>O<sub>2</sub>). Similar acetoxonium ion intermediates have been reported in the rearrangements of oxiranes.<sup>10,11</sup>

Since the rearrangement of the 6 $\beta$ -acetoxy-3 $\alpha$ ,5 $\alpha$ -oxetane (2) proceeded quite smoothly, we suspected the structural assignments of the 3 $\beta$ ,5 $\beta$ -oxetanes (7)<sup>5</sup> and (8). A study of models and comparison of the spectroscopic data (Table) for the 3 $\beta$ ,5 $\beta$ -oxetan-6-one (7) and the 3 $\alpha$ ,5 $\alpha$ -oxetan-6-one (4) which was prepared from (3) by oxidation, led to the conclusion that the solvolysis product (7) should be formulated as the A-homo-B-nor-3 $\alpha$ ,5 $\alpha$ -epoxide (18). This could arise by the rearrangement<sup>12</sup> shown in Scheme 2 which would allow normal intramolecular concerted displacement of the tosyloxy-group.

Table

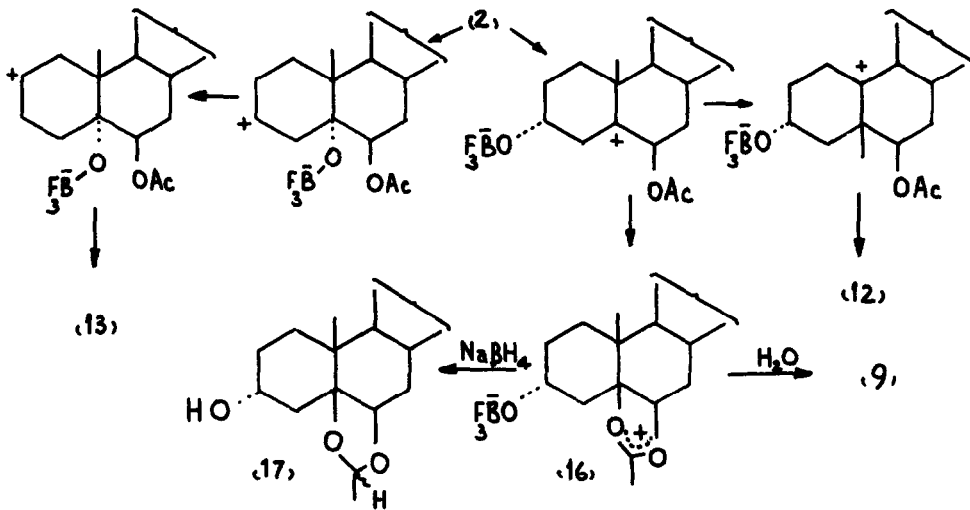
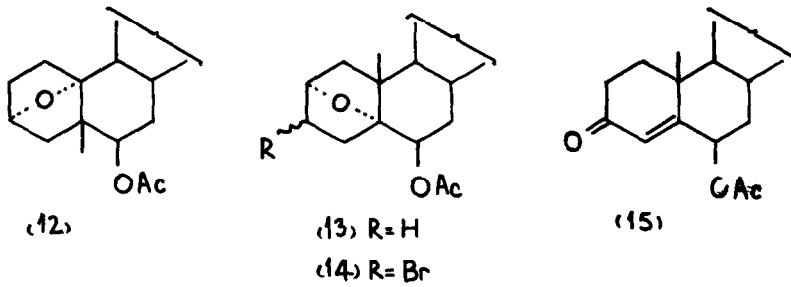
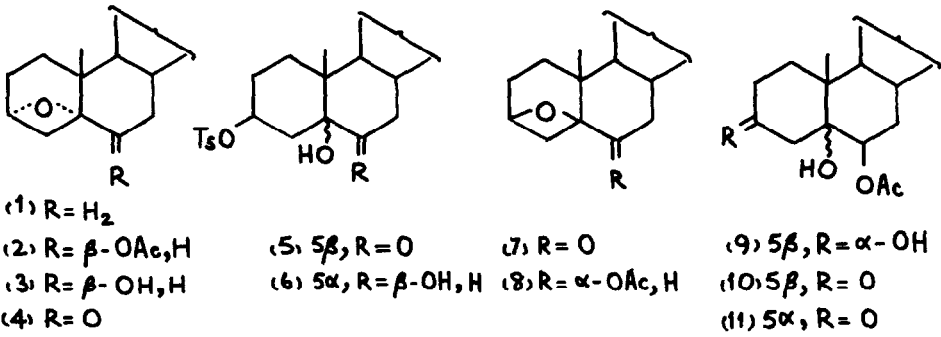
Compound	$\delta$		$\nu_{\max}$ cm <sup>-1</sup>
	3-H	4-H	
4	4.45, d(br), J <sub>v</sub> 7 Hz	$\alpha$ 3.0, q, J <sub>v</sub> 10 and 7 Hz; $\beta$ 1.86, d, J <sub>v</sub> 10 Hz	1723
7/18	4.67, t(br), J <sub>v</sub> 7 Hz	2.62, q, J <sub>v</sub> 18 and 7 Hz; 2.08, d, J <sub>v</sub> 18 Hz	1750

NaBH<sub>4</sub> reduction of the solvolysis product (18) gave a single alcohol (20)<sup>2</sup> and, as would be expected, the 10-Me signal in the <sup>1</sup>H n.m.r. spectrum was considerably downfield ( $\delta$ 1.05) from that in the parent ketone ( $\delta$ 0.73). Further support for the structure (18) was indicated by deuteration in Et<sub>3</sub>N/D<sub>2</sub>O/dioxan. Even after several treatments, the 4 $\alpha$ -monodeuterio-analogue (19) was largely

<sup>†</sup> Kindly provided by Professor E. Glotter.

\*\* It is not known to what extent the subsequent rearrangements are concerted with these initial cleavages.

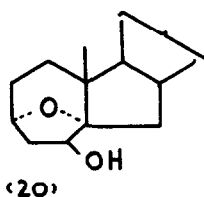
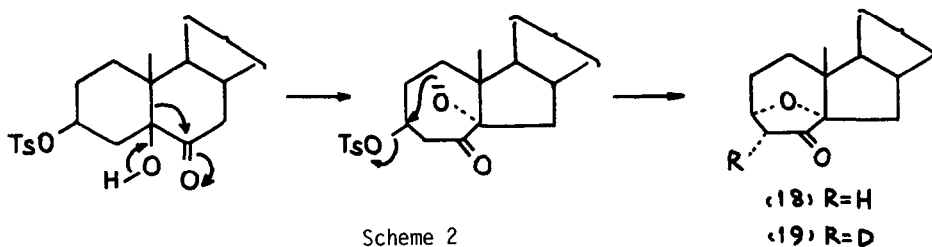
§ We are extremely grateful to Professor A.T. Rowland for a generous sample.



Scheme 1

obtained as indicated by the mass spectrum and by the  $^1\text{H}$  n.m.r. spectrum in which the 3-H signal was a broadened doublet (major residual coupling to  $2\alpha\text{-H}$ ) and the  $4\beta\text{-H}$  signal ( $\delta 2.08$ ) was a broadened singlet. As expected the signal for the  $4\alpha\text{-H}$  ( $\delta 2.62$ ) had disappeared. The reluctance of the  $4\beta\text{-H}$  to exchange significantly presumably relates to the hindered nature of the base and of the  $\beta$ -face of the 4,4a-enolate. The loss of CO and  $\text{C}_4\text{H}_5\text{D}$  in the mass spectrum of (19) gives the base peak at  $m/e$  318 and further supports the structural assignment.

During the preparation of this manuscript an independent correction of the structure (7) has been reported.<sup>13</sup>



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